Processing The Interspecies Quorum-Sensing Signal Autoinducer-2 (AI-2) Characterization Of Phospho-(S)-4,5-Dihydroxy-2,3-Pentanedione Isomerization By LsrG Protein

J. C. Marques

P. Lamosa

Caitlin Elizabeth Russell , '11

R. Ventura

C. D. Maycock

See next page for additional authors

Follow this and additional works at: https://works.swarthmore.edu/fac-chemistry

Part of the Biochemistry Commons

Let us know how access to these works benefits you

Recommended Citation


https://works.swarthmore.edu/fac-chemistry/54

This work is brought to you for free by Swarthmore College Libraries' Works. It has been accepted for inclusion in Chemistry & Biochemistry Faculty Works by an authorized administrator of Works. For more information, please contact myworks@swarthmore.edu.
Authors

J. C. Marques; P. Lamosa; Caitlin Elizabeth Russell , ’11; R. Ventura; C. D. Maycock; M. F. Semmelhack; Stephen T. Miller; and K. B. Xavier

This article is available at Works: https://works.swarthmore.edu/fac-chemistry/54
Processing the Interspecies Quorum-sensing Signal Autoinducer-2 (AI-2)

CHARACTERIZATION OF PHOSPHO-(S)-4,5-DIHYDROXY-2,3-PENTANEDIONE ISOMERIZATION BY LsrG PROTEIN*

Received for publication, February 12, 2011, and in revised form, March 29, 2011. Published, JBC Papers in Press, March 30, 2011, DOI 10.1074/jbc.M111.230227

João C. Marques†, Pedro Lamosa‡, Caitlin Russell§, Rita Ventura‡, Christopher Maycock†, Martin F. Semmelhack**, Stephen T. Miller†, and Karina B. Xavier†,‡

From the †Instituto Gulbenkian de Ciência and the ‡Instituto de Tecnologia Química e Biológica, 2780 Oeiras, Portugal, the ¶Department of Chemistry and Biochemistry, Swarthmore College, Swarthmore, Pennsylvania 19081, the †Faculdade de Ciências da Universidade de Lisboa, Departamento de Química e Bioquímica, 1749 Lisboa, Portugal, and the **Department of Chemistry, Princeton University, Princeton, New Jersey 08544

The molecule (S)-4,5-dihydroxy-2,3-pentanedione (DPD) is produced by many different species of bacteria and is the precursor of the signal molecule autoinducer-2 (AI-2). AI-2 mediates interspecies communication and facilitates regulation of bacterial behaviors such as biofilm formation and virulence. A variety of bacterial species have the ability to sequester and process the AI-2 present in their environment, thereby interfering with the cell-to-cell communication of other bacteria. This process involves the AI-2-regulated lsr operon, comprised of the Lsr transport system that facilitates uptake of the signal, a kinase that phosphorylates the signal to phospho-DPD (P-DPD), and enzymes (like LsrG) that are responsible for processing the phosphorylated signal. Because P-DPD is the intracellular inducer of the lsr operon, enzymes involved in P-DPD processing impact the levels of Lsr expression. Here we show that LsrG catalyzes isomerization of P-DPD into 3,4,4-trihydroxy-2-pentanone-5-phosphate. We present the crystal structure of LsrG, identify potential catalytic residues, and determine which of these residues affects P-DPD processing in vivo and in vitro. We also show that an lsrG deletion mutant accumulates at least 10 times more P-DPD than wild type cells. Consistent with this result, we find that the lsrG mutant has increased expression of the lsr operon and an altered profile of AI-2 accumulation and removal. Understanding of the biochemical mechanisms employed by bacteria to quench signaling of other species can be of great utility in the development of therapies to control bacterial behavior.

Many bacteria regulate gene expression as a function of the density of the population. This process, called quorum sensing, enables these organisms to coordinate behaviors that are most beneficial when cells are working in unison (1, 2). Quorum sensing is mediated by signal molecules called autoinducers. One autoinducer (Autoinducer-2 (AI-2)) is produced by many species of bacteria and can facilitate interspecies cell-cell signaling (3–5). AI-2 is produced by the enzyme LuxS, which synthesizes (S)-4,5-dihydroxy-2,3-pentanedione (DPD; Fig. 1A, 1), the linear form of a set of interconverting molecules with AI-2 activity (6–8). AI-2 (or its synthase LuxS) has been shown to regulate important bacterial behaviors such as biofilm formation and the production of virulence factors (4, 5, 9, 10). Thus, one strategy for controlling these behaviors is to control the concentration (or availability) of AI-2. We have previously shown that certain bacteria can employ this strategy by processing AI-2 and thus quenching interspecies signaling (11). These bacteria include most members from the Enterobacteriaceae family (like the commensal Escherichia coli K12 and the pathogen E. coli O157 and Salmonella typhimurium) but also more distantly related bacteria such as the plant symbiont Sinorhizobium meliloti and the pathogen Bacillus anthracis (12).

In these bacteria, AI-2 controls the expression of a system (named Lsr for luxS-regulated) capable of processing both endogenous and exogenous AI-2 (13–16). Using the Lsr system, these bacteria remove AI-2 from the environment and, thus, eliminate the ability of others to utilize AI-2 to regulate their behaviors (11). Upon internalization by the Lsr transporter, AI-2 is phosphorylated to phospho-DPD (P-DPD, Fig. 1A, 2) by the kinase LsrK (17, 18). P-DPD is the inducer of the Lsr system and acts by binding to and inactivating the transcriptional repressor LsrR (18, 19). Thus, P-DPD triggers the induction of the Lsr system, causing a fast uptake of extracellular AI-2 and processing of internalized AI-2. Therefore, it is expected that

*This work was supported, in whole or in part, by National Institutes of Health Grant of AI074041 (to S. T. M.). This work was also supported by the Fundação para a Ciência e Tecnologia (PPCDT/ DG/BIA/82010/2006, Portugal (to K. B. X.)) and by the Camille and Henry Dreyfus Foundation and Swarthmore College (to S. T. M.).

†The on-line version of this article (available at http://www.jbc.org) contains supplemental Figs. S1–S7.

The molecular co factors (code 3QM) have been deposited in the Protein Data Bank, Research Collaboratory for Structural Bioinformatics, Rutgers University, New Brunswick, NJ (http://www.rcsb.org/).

1To whom correspondence may be addressed: 500 College Ave., Swarthmore, PA 19081. Tel.: 610-957-0663; Fax: 610-328-7355; E-mail: smiller1@swarthmore.edu.

2To whom correspondence may be addressed: R. da Quinta Grande 6, 2780-156 Oeiras, Portugal. Tel.: 351-21-4464655; Fax: 351-21-4407970; E-mail: kxavier@igc.gulbenkian.pt.

3The abbreviations used are: AI-2, autoinducer-2; DPD, (S)-4,5-dihydroxy-2,3-pentanedione; P-DPD, phospho-DPD; Lsr, luxS regulated; P-TPO, 3,4,4-trihydroxy-2-pentanone-5-phosphate; P-HPD, 3-hydroxy-2,4-pentadione-5-phosphate; P-TetraPO, 3,3,4,4-tetrahydroxy-2-pentanone-5-phosphate; HSOQ, heteronuclear single quantum coherence; HMBC, heteronuclear multiple bond connectivity; Cm, chloramphenicol; r.m.s.d., root mean square deviation.

JOURNAL OF BIOLOGICAL CHEMISTRY 18331
Isomerization of P-DPD by LsrG

![Model of the proposed reaction pathway for P-DPD isomerization by LsrG.](http://www.jbc.org/)

A, in the presence of ATP and reducing agent DPD (1) is phosphorylated by LsrK, producing phospho-DPD (P-DPD, 2). The hydrated form of P-DPD is also shown because this is the major form of P-DPD detected under our reaction conditions. LsrG converts P-DPD into 3,4,4-trihydroxy-2-pentanone-5-phosphate (P-TPO, 3) by catalyzing the tautomerization of P-DPD into 3-hydroxy-2,4-pentadione-5-phosphate (P-HPD, 4). For clarity and easy description of structural differences, the numbering of the carbons of all the compounds is based on the numbering of DPD.

B, in the presence of oxygen P-TPO is oxidized non-enzymatically to 3,3,4,4-tetrahydroxy-2-pentanone-5-phosphate (P-TetraPO, 5). For clarity and easy description of structural differences, the numbering of the carbons of all the compounds is based on the numbering of DPD.

FIGURE 1. Model of the proposed reaction pathway for P-DPD isomerization by LsrG. A, in the presence of ATP and reducing agent DPD (1) is phosphorylated by LsrK, producing phospho-DPD (P-DPD, 2). The hydrated form of P-DPD is also shown because this is the major form of P-DPD detected under our reaction conditions. LsrG converts P-DPD into 3,4,4-trihydroxy-2-pentanone-5-phosphate (P-TPO, 3) by catalyzing the tautomerization of P-DPD into 3-hydroxy-2,4-pentadione-5-phosphate (P-HPD, 4) via a transient 3,4-enediol intermediate and subsequent hydration of the carbon 4-ketone group. It is likely that the hydrated (3) and non-hydrated (4) forms exist in equilibrium. B, in the presence of oxygen P-TPO is oxidized non-enzymatically to 3,3,4,4-tetrahydroxy-2-pentanone-5-phosphate (P-TetraPO, 5). For clarity and easy description of structural differences, the numbering of the carbons of all the compounds is based on the numbering of DPD.

Every enzyme acting on P-DPD has a significant impact on AI-2 internalization and, hence, on Al-2-dependent interspecies cell-cell signaling. We have previously shown that LsrG is one of the enzymes involved in processing of P-DPD (18). We showed that under aerobic conditions, in vitro incubation of LsrG with P-DPD results in the production of phosphoglycolate as well as multiple other compounds that accumulate to low levels. Neither the catalytic mechanism of LsrG nor the other compounds produced in the LsrG reaction were identified.

Here, we characterize the enzymatic reaction catalyzed by LsrG. We show that LsrG catalyzes the conversion of P-DPD to the isomer 3,4,4-trihydroxy-2-pentanone-5-phosphate (P-TPO, Fig. 1A, 3) and propose the reaction that leads to the formation of this compound (Fig. 1A). Under reducing conditions (i.e. the conditions typically found in the cytoplasm) P-TPO and the non-hydrated form of this compound (3-hydroxy-2,4-pentadione-5-phosphate (P-HPD, Fig. 1A, 4) are stable and do not degrade readily. However, under oxidizing conditions, P-TPO and P-HPD are oxidized to a compound we identify as 3,3,4,4-tetrahydroxy-2-pentanone-5-phosphate (P-TetraPO, Fig. 1B, 5), which in turn degrades into phosphoglycolate and other smaller molecules produced in minor concentrations.

We also present the crystal structure of LsrG. LsrG shares a fold with proteins belonging to the cofactor-independent monooxygenase family (20, 21). Interestingly, although the location of the LsrG active site is the same as in monooxygenases, the identity of the residues in the active site is not highly conserved. The LsrG structure has allowed us to identify potential catalytic residues, identifications that were then validated in vitro and in vivo. Finally, to determine the role of LsrG in the process of interference with AI-2 mediated signaling, we measured the impact of LsrG mutants on the intracellular accumulation of P-DPD, on the expression of the lsr operon, and on extracellular AI-2 accumulation/removal. Characterization of LsrG and the role it plays in AI-2 processing is essential for understanding how organisms that possess the Lsr system act to quench AI-2 signaling. This knowledge opens the door to the potential exploitation of these mechanisms as a means for controlling quorum sensing.

**EXPERIMENTAL PROCEDURES**

**Overexpression and Purification of LsrG**

*E. coli* LsrG was cloned into the pProEXHTb plasmid for overexpression as an N-terminal His₆-tagged fusion protein. After transformation of the plasmid into *E. coli* strain BL21, cultures were grown in Luria-Bertani medium (Sigma) with shaking at 37 °C to an A₅₉₅ of 0.3. The temperature was lowered to 22 °C, and growth continued until an A₅₉₅ of 0.9 was reached. Protein expression was induced by the addition of 0.1 mM isopropyl-β-D-thiogalactopyranoside. After 15 h of growth at 22 °C with shaking post-induction, bacteria were harvested by centrifugation. Cells were resuspended in 50 mM NaH₂PO₄, pH 8.0, 300 mM NaCl, 10 mM imidazole, 1.4 mM β-mercaptoethanol, 2.5 μg ml⁻¹ DNase, 2.5 μg ml⁻¹ leupeptin, and 2.5 μg ml⁻¹ aprotinin. Cells were lysed using a M-110Y Microfluidizer (Microfluidics). Lysates were clarified via centrifugation.

The His₆-LsrG fusion protein was purified by affinity chromatography using nickel-nitritetriacetic acid-agarose (Qiagen). After application of the lysate, the columns were washed with 50 mM NaH₂PO₄, pH 8.0, 300 mM NaCl, 20 mM imidazole, and 1.4 mM β-mercaptoethanol. The fusion protein was eluted from the columns using 50 mM NaH₂PO₄, pH 8.0, 300 mM NaCl, 250 mM imidazole, and 1.4 mM β-mercaptoethanol. Protein-con-
Diffusion of P-DPD by LsrG

DPD Synthesis and Preparation

DPD protected with a cyclohexylidene group was chemically synthesized by the recently developed method as reported in Ascenso et al. (22) and dissolved in water to 10 mM. The cyclohexylidene-protecting group was removed with acidic Dowex 50WX8 resin (100 mg ml⁻¹ of sample) also as reported previously (22). Deprotected DPD was then neutralized with a final concentration of 30 mM potassium phosphate buffer, pH 7.2. The released cyclohexanone was removed by liquid-liquid extraction with an equal volume of deuterated chloroform. Removal of cyclohexanone is essential to facilitate assignment of the resonances in the methyl region of the NMR spectra. To determine the final concentration of DPD, quantification was performed by ¹H NMR using formate as an internal concentration standard.

NMR Spectroscopy

Unless otherwise stated, all spectra were acquired on a Bruker AVANCE III 800 spectrometer (Bruker, Rheinstetten, Germany) working at a proton operating frequency of 800.33 MHz, equipped with a four-channel 5-mm inverse detection probe head with pulse-field gradients along the Z axis. Spectra were run at 4 °C (to avoid compound degradation) using standard Bruker pulse programs. ¹H and ¹³C chemical shifts were referenced to 3-(trimethylsilyl)propionate sulfonic acid. ³¹P chemical shifts were referenced to external 85% phosphoric acid. Standard Bruker pulse programs were used to acquire the heteronuclear ¹³C/¹H two-dimensional correlation spectra. In the ¹³C/¹H heteronuclear single quantum coherence (HSQC) spectrum, a delay of 3.45 ms was used for evolution of ¹H/³¹P whereas a delay of 62.5 ms was used for the ¹H/³¹P correlation spectra. In the heteronuclear multiple bond connectivity (HMBC) spectrum a delay of 73.5 ms was used for evolution of long range couplings. In the HSQC type spectra, proton decoupling was achieved using the GARP4 sequence (23). For the enzyme kinetics determinations, spectra were acquired on a Bruker AVANCE II 500 spectrometer (Bruker) equipped with a 5-mm broadband inverse detection probe head, working at 30 °C with a proton operating frequency of 500.43 MHz. Typically, ¹H NMR spectra were acquired with water presaturation using a 60° flip angle and a repetition delay of 5.23 s. The same conditions were used to analyze the cell extracts, but spectra were collected at 4 °C. For quantification purposes formate was added as an internal concentration standard, and an extra relaxation delay of 30 s was used. All samples contained a mixture of D₂O/H₂O with 5% D₂O. For the identification of P-HPD, the correlation spectra were performed in 100% D₂O; in this case, the LsrG reaction was performed with P-DPD that had been lyophilized and was suspended in D₂O before the addition of LsrG.

In Vitro Enzyme Assays

LsrK Reaction and Preparation of P-DPD—P-DPD was produced enzymatically by the phosphorylation of DPD with purified LsrK and ATP. LsrK was purified as described previously (18). The reaction mixture contained 30 µg ml⁻¹ LsrK, 2–3 mM DPD, 6 mM ATP, 24 mM MgCl₂, 100 mM potassium phosphate buffer pH 7.2, and 5% D₂O. Reaction mixtures were incubated at 30 °C for 10 min and terminated by placing the samples in ice. Typically, these conditions ensured a complete conversion of DPD to P-DPD, and this was always confirmed by ¹H NMR or ³¹P NMR spectra (at 4 °C to avoid degradation).

LsrG Assays—To analyze the products of the LsrG reaction, 3 µg ml⁻¹ final concentration of purified enzyme was added to the reaction mixture containing enzymatically prepared P-DPD. To ensure reducing conditions, samples were degassed with argon for 5 min, and freshly prepared DTT was added to a final concentration of 10 mM (unless mentioned otherwise) before the addition of LsrG. Incubation of these mixtures for 2 min at 30 °C was sufficient to ensure complete consumption of P-DPD. In cases where the LsrG reaction was performed in anaerobiosis, the LsrK reaction products were degassed by three cycles of vacuum and argon, and the LsrG enzyme and the diethylpyrocarbonate solution were left in an inert atmosphere glove box for 20 min. Mixtures were created in anaerobic conditions. To inhibit LsrG, diethylpyrocarbonate was used at a final concentration of 50 mM. The enzymatic activity of LsrG and its mutants was determined by monitoring the kinetics of P-DPD decrease at 30 °C by acquiring sequential ¹H NMR spectra (1.5 min each). For these experiments a batch of P-DPD was prepared as detailed above and quantified by ¹H NMR using formate as an internal reference. This mixture was kept in ice until use. Before each reaction an aliquot of 650 µl of the P-DPD sample was incubated for 5 min at 30 °C in a 5-mm NMR tube, and the reaction was started by adding the LsrG or its mutants.
Isomerization of P-DPD by LsrG

and placed in the spectrometer. The first spectra were started 1.5 min after enzyme addition. Controls with no enzyme or no substrate were performed for all conditions. To determine the specific enzyme activity, the concentration of protein added was adjusted to ensure that the rate of substrate consumption was linear at least in the first 10–12 min. For wild type (WT) LsrG, 1 μg ml⁻¹ of protein was used, but for the point mutants, concentrations up to 20 μg ml⁻¹ were tested.

**Crystallization and Structure Determination**

Crystals of WT LsrG with the His₆ tag removed were grown via the sitting drop method with a well solution of 1.65 M sodium citrate, pH 6.5. Crystals were cryoprotected in paratone and flash-frozen in the diffractometer cryostream. Data were collected at National Synchrotron Light Source beamline X25. The crystals (P2₁, a = 44.64 Å, b = 83.39 Å, c = 63.95 Å, β = 89.52°) diffracted to 1.8 Å resolution. Data were processed using Denzo, Scalepack, and CCP4 (24, 25).

The crystals of LsrG were determined using molecular replacement with PHENIX (26) using LsrG from *Yersinia pestis* (Protein Data Bank ID 2GFF) as the search model. The model was rebuilt using Coot (27) and refined with PHENIX (26) and REFMAC (28). The model exhibits good geometry (Table 1), with no residues outside the allowed region of the Ramachandran plot. The final model includes 288 water molecules and has a final R cryst of 0.204 and R free of 0.240. Coordinates and structure factors for LsrG were deposited in the Protein Data Bank with accession number 3QMQ.

Structural alignments were calculated in Coot (27), and sequence alignments were calculated by ClustalW2 (29). All molecular images were generated using PyMOL (30).

**Bacterial Strains, Plasmids, and Growth Conditions**

All strains and plasmids used are listed in Table 2. These strains are all *E. coli* MG1655 derivatives of KX1123 (ΔlsrZYA, lsr-lacZ (14)). The strain JCM16 ΔlsrG::Cm was constructed by replacing the lsrG gene in KX1123 by a chloramphenicol (Cm) resistance cassette as described previously (31) using primers with 50 bp of homology to the flanking regions of lsrG and was introduced in KX1123 by generalized transduction with bacteriophage P1 (32). To construct JCM23, Cm was removed from strain JCM16 with pcp20 as described (31). To obtain JCM62 (ΔlsrK::Kan) a P1 lysate from the strain JW1504 ΔlsrK::Kan (obtained from National BioResource Project (Japan): *E. coli*) was used to introduce this deletion into KX1123 by generalized P1 transduction. All plasmids were transformed by electroporation according to standard protocols (33). Except where otherwise stated, strains were grown in Luria-Bertani medium supplemented with 100 mM MOPS, pH 7, at 37 °C with shaking. Growth was monitored by optical density at 595 nm (A₅₉₅).

**In Vivo Expression of Transcription of the lsr Operon**

Expression of lsr transcription in *E. coli* was measured by determining the β-galactosidase activity of the lsr-lacZ promoter fusion as described before (14). To test the effect of the expression of LsrG and its mutants, the strains containing the pProExHtb plasmid and its derivatives were cultured in Luria-Bertani medium supplemented with 100 mM MOPS, pH 7.0, ampicillin (100 mg liter⁻¹) and isopropyl 1-thio-D-galactopyranoside (1 μg liter⁻¹). Isopropyl 1-thio-β-D-galactopyranoside was added 2.5 h after inoculation to mimic the start of lsrG induction in the WT strain. To assay β-galactosidase activity,

---

**TABLE 1**

<table>
<thead>
<tr>
<th>Chrystallographic data and refinement statistics</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Data collection</strong></td>
</tr>
<tr>
<td>Resolution (Å)</td>
</tr>
<tr>
<td>Unique reflections</td>
</tr>
<tr>
<td>Rmerge</td>
</tr>
<tr>
<td>Mean I/σI</td>
</tr>
<tr>
<td>Completeness (%)</td>
</tr>
<tr>
<td>Multiplicity</td>
</tr>
</tbody>
</table>

**Refinement**

Rmerge/Rfree | 0.204/0.243 |
| r.m.s.d. bond length (Å) | 0.023 |
| r.m.s.d. bond angle (°) | 1.97 |
| No. of atoms per asymmetric unit | 3,508 |

**Ramachandran plot**

Most favored (%) | 95.5 |
| Allowed (%) | 4.5 |
| Disallowed (%) | 0.0 |

* The highest resolution shell is shown in parenthesis.

**TABLE 2**

<table>
<thead>
<tr>
<th>E. coli strains and plasmids used in this study</th>
<th>Relevant genotype</th>
<th>Source</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>E. coli strains</strong></td>
<td></td>
<td></td>
</tr>
<tr>
<td>KX1123</td>
<td>lsr-lacZ, ΔlsrZYA</td>
<td>Xavier and Bassler (14)</td>
</tr>
<tr>
<td>KX1186</td>
<td>lsr-lacZ, ΔlsrK::Tn10Cm</td>
<td>Xavier and Bassler (14)</td>
</tr>
<tr>
<td>JCM16</td>
<td>lsr-lacZ, ΔlsrG::Cm</td>
<td>This study</td>
</tr>
<tr>
<td>JCM23</td>
<td>lsr-lacZ, ΔlsrG</td>
<td>This study</td>
</tr>
<tr>
<td>JCM43</td>
<td>lsr-lacZ, ΔlsrG::Cm, pLsrG-WT</td>
<td>This study</td>
</tr>
<tr>
<td>JCM44</td>
<td>lsr-lacZ, ΔlsrG::Cm, pLsrG-E54A</td>
<td>This study</td>
</tr>
<tr>
<td>JCM45</td>
<td>lsr-lacZ, ΔlsrG::Cm, pLsrG-H65A</td>
<td>This study</td>
</tr>
<tr>
<td>JCM46</td>
<td>lsr-lacZ, ΔlsrG::Cm, pLsrG-N25A</td>
<td>This study</td>
</tr>
<tr>
<td>JCM47</td>
<td>lsr-lacZ, ΔlsrG::Cm, pProExHtb</td>
<td>This study</td>
</tr>
<tr>
<td>JCM48</td>
<td>lsr-lacZ, ΔlsrG::Cm, pLsrG-H70A</td>
<td>This study</td>
</tr>
<tr>
<td>JCM53</td>
<td>lsr-lacZ, pProExHtb</td>
<td>This study</td>
</tr>
<tr>
<td>JCM62</td>
<td>lsr-lacZ, ΔlsrK::Kan, pProExHtb</td>
<td>This study</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Plasmids</th>
<th></th>
<th></th>
</tr>
</thead>
<tbody>
<tr>
<td>pProExHtb</td>
<td>Expression vector with hexahistidine tag (Amp⁺)</td>
<td></td>
</tr>
<tr>
<td>pLsrG-WT</td>
<td>pProExHtb containing lsrG-WT</td>
<td></td>
</tr>
<tr>
<td>pLsrG-E54A</td>
<td>pProExHtb containing lsrG-E54A</td>
<td></td>
</tr>
<tr>
<td>pLsrG-H65A</td>
<td>pProExHtb containing lsrG-H65A</td>
<td></td>
</tr>
<tr>
<td>pLsrG-N25A</td>
<td>pProExHtb containing lsrG-N25A</td>
<td></td>
</tr>
<tr>
<td>pLsrG-H70A</td>
<td>pProExHtb containing lsrG-H70A</td>
<td></td>
</tr>
</tbody>
</table>
cells were collected and harvested when cultures reached $A_{595} = 3$. To measure $lsr$-lacZ expression in cell cultures during growth, extracts were created from 1 ml of culture aliquots collected in triplicate at the indicated times. All assays were performed in triplicate.

**Time Course of Al-2 Production in Bacterial Cultures**

To monitor extracellular Al-2 activity in cell cultures during growth, aliquots were collected at the indicated times and cell-free culture fluids were prepared by the filtration of liquid cultures (34, 35) and then analyzed for Al-2 activity. Quantification of Al-2 was performed using the *in vitro* assay based on the LuxP-FRET method established by Rajamani *et al.* (36) and optimized for 96-well plate reading using a multilabel counter (1420 Victor 3, PerkinElmer Life Sciences). Specifically, serial dilutions of sample were added to 12.5 μg ml⁻¹ CFP-LuxP-YFP chimeric protein in 25 mM sodium phosphate buffer, pH 8, 35 mM NaCl, and 1 mM borate. Samples (2.5 μl) were added to a 280-μl reaction volume, and Al-2 concentrations of the samples were calculated from the FRET ratio (527/485 nm). Binding of Al-2 to the CFP-LuxP-YFP protein causes a dose-dependent decrease in the FRET signal, and concentration can be determined by comparing the FRET ratios of each sample with a calibration curve performed with Al-2 samples of known concentration produced and quantified as described in Schauder *et al.* (6). Concentrations between 1 and 60 μM Al-2 were used for the calibration curve, corresponding to the linear range of this assay. All assays were performed in triplicate.

**Identification and Quantification of P-PDP in Cell Extracts**

Each strain was cultured to $A_{595} = 3$ (conditions at which expression of $lsr$ expression is high) and harvested by centrifugation at 7000 × g for 10 min at 4 °C. Cells were washed twice with 100 mM potassium phosphate buffer, pH 7.2, containing 1 mM MgCl₂; the cell pellets were resuspended in the same buffer and incubated with 2 mM DPD. After the addition of DPD, the cell suspension had a final volume of 3 ml and a calculated $A_{595}$ of 70–80. This mixture was incubated at 30 °C with aeration. Efficient mixing and supply of oxygen to the cell suspension were achieved by bubbling oxygen and using an airlift system (37). After 15 min of incubation, ice-cold perchloric acid and DTT (0.6 M and 10 mM final concentrations, respectively) were added. The mixture was incubated for 20 min with stirring in an ice bath. 5 M KOH was then added to achieve pH 5. This extract was centrifuged for 15 min at 9000 × g at 4 °C. 5% D₂O, 5 mM EDTA, and 400 mM formate were added to the extracts for ¹H NMR analysis at 4 °C as described above.

**RESULTS**

*LsrG Catalyzes Conversion of P-DPD to the Isomer P-TPO*—When LsrG is incubated with P-DPD under reducing (i.e. in the presence of DTT) or anaerobic conditions, P-DPD is completely converted to two phosphorylated compounds as shown by ³¹P NMR spectra (Fig. 2). The resonance at 4.75 ppm was assigned to the P-DPD isomer, P-TPO (Fig. 1, 3) and the resonance with smaller intensity at 3.73 ppm was assigned to P-HPD (Fig. 1, 4), the non-hydrated form of P-TPO. These compounds were identified by NMR spectroscopy, and the chemical shifts of the assigned resonances are summarized on Table 3. The ³¹P resonances of the LsrG products are typical of phosphomonoester compounds, and their proton coupling patterns are triplets with coupling constants of 8.5 and 6.4 Hz for P-TPO and P-HPD, respectively. This pattern revealed that these two phosphate groups are linked to methylene groups. ¹H,³¹P correlation experiments provided the chemical shifts of the resonances of the protons from the methylene groups asso-
Isomerization of P-DPD by LsrG

TABLE 3

<table>
<thead>
<tr>
<th>NMR parameters of P-DPD, P-TPO, P-HPD, P-TetraPO, and PG</th>
</tr>
</thead>
</table>

<table>
<thead>
<tr>
<th>13C NMR δ (in ppm)</th>
<th>C-1</th>
<th>C-2</th>
<th>C-3</th>
<th>C-4</th>
<th>C-5</th>
</tr>
</thead>
<tbody>
<tr>
<td>P-DPD (2)</td>
<td>27.2</td>
<td>212.6</td>
<td>99.2</td>
<td>75.6</td>
<td>66.7</td>
</tr>
<tr>
<td>P-TPO (3)</td>
<td>31.4</td>
<td>216.0</td>
<td>81.0</td>
<td>98.6</td>
<td>69.1</td>
</tr>
<tr>
<td>P-HPD (4)</td>
<td>29.3</td>
<td>210.0</td>
<td>84.4</td>
<td>208.4</td>
<td>70.5</td>
</tr>
<tr>
<td>P-TetraPO (5)</td>
<td>29.2</td>
<td>213.3</td>
<td>99.9</td>
<td>98.2</td>
<td>68.4</td>
</tr>
<tr>
<td>PCb</td>
<td>181.0</td>
<td>66.1</td>
<td>-</td>
<td>-</td>
<td>-</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>31P NMR δ (in ppm)</th>
<th>Phosphate</th>
</tr>
</thead>
<tbody>
<tr>
<td>P-DPD (2)</td>
<td>3.90</td>
</tr>
<tr>
<td>P-TPO (3)</td>
<td>4.75</td>
</tr>
<tr>
<td>P-HPD (4)</td>
<td>3.73</td>
</tr>
<tr>
<td>P-TetraPO (5)</td>
<td>4.82</td>
</tr>
<tr>
<td>PG</td>
<td>2.88</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>1H-NMRδ (in ppm)</th>
<th>H-1</th>
<th>H-2</th>
<th>H-3</th>
<th>H-4</th>
<th>H-5a</th>
<th>H-5b</th>
</tr>
</thead>
<tbody>
<tr>
<td>P-DPD (2)</td>
<td>2.37 (s)</td>
<td>-</td>
<td>-</td>
<td>4.11 (dd, 3.68, 6.93)</td>
<td>4.02 (m)</td>
<td>3.88 (dt, 11.34, 6.74)</td>
</tr>
<tr>
<td>P-TPO (3)</td>
<td>2.38 (s)</td>
<td>-</td>
<td>4.37 (s)</td>
<td>-</td>
<td>3.89 (t, 1.028)</td>
<td>3.76 (dd, 7.07, 11.19)</td>
</tr>
<tr>
<td>P-HPD (4)</td>
<td>2.35 (s)</td>
<td>-</td>
<td>-</td>
<td>4.78 (m)</td>
<td>4.64 (m)</td>
<td></td>
</tr>
<tr>
<td>P-TetraPO (5)</td>
<td>2.44 (s)</td>
<td>-</td>
<td>-</td>
<td>3.92 (d, 8.79)</td>
<td>3.92 (d, 8.79)</td>
<td></td>
</tr>
<tr>
<td>PG</td>
<td>-</td>
<td>4.17 (d, 5.21)</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td></td>
</tr>
</tbody>
</table>

a This chemical shift shows that the P-DPD detected is hydrated at position 3.
b PC, phosphoglycolate.
c Multiplicities and coupling constants (Hz) are given in parentheses.
d ND, Not detected.

ciated with these phosphorous groups (Table 3, and supplemental Fig. S1, lower panel). Next we acquired a set of 12C, 1H HMBC and 13C, 1H HSQC NMR spectra to identify the other groups of these two compounds. In the assignment of P-TPO, the 13C, 1H HMBC experiment (which correlates proton signals with neighboring carbon resonances two or three bonds apart) revealed that the protons from the phosphorylated carbon 5 of P-TPO, identified in the 1H, 31P correlation experiments and labeled as 3H5a and 3H5b (Fig. 3) correlated with two carbon signals. Specifically, the chemical shift of the most intense correlation (at 98.6 ppm), assigned to carbon 4 (Fig. 3, 3H5a/b → 3C4), is consistent with a gem-diol, whereas the weaker correlation (at 81.0 ppm), assigned to carbon 3 (Fig. 3, 3H5a/b → 3C3), is typical of an alcohol group. We then used the 13C, 1H HSQC spectrum (supplemental Fig. S2), which determines direct correlation between carbons and attached protons to determine the proton signal of carbon 3 from P-TPO (4.37 ppm). Again, in the 13C, 1H HMBC spectrum, the proton bound to carbon 3 (Fig. 3, 3H5a/b → 3C3) revealed a correlation with carbon 4 (identified above) and another at 216 ppm, identified as a carbonyl group assigned as carbon 2 (Fig. 3, 3H5a/b → 3C2). This resonance from carbon 2 in turn is correlated with a proton resonance at 2.38 ppm in the methyl region (Fig. 3, 3H5a/b → 3C2), which in the HSQC spectrum showed a correlation with a carbon at 31.4 pm assigned as carbon 1 (supplemental Fig. S2, Table 3). Overall, these data lead us to propose that the main product of the LsrG reaction (Fig. 1) is the compound P-TPO.

1H NMR spectra of the reaction mixture, resulting from incubation of P-DPD with LsrG under reducing conditions, show two resonances in the methyl region: one at 2.38 ppm assigned to P-TPO and a second one at 2.35 ppm with lower intensity (supplemental Fig. S3, middle trace). This result is consistent with the phosphorous spectrum mentioned above (Fig. 2). Given that the proton resonance at 2.35 ppm is present only when the weaker phosphorous resonance labeled as P-HPD in Fig. 2 is also present, we conclude that this methyl group and the phosphate group belong to the same compound, a conclusion supported by the relative intensities of the signals in the phosphorous and proton spectra (Fig. 2 and supplemental Fig. S3). Using a combination of 13C, 1H HMBC (Fig. 3 and supplemental Fig. S1) and 13C, 1H HSQC (supplemental Fig. S2) correlation spectra and following the strategy used for the assignment of P-TPO, we determined all the carbon resonances of this second compound (Table 3), allowing us to propose that this second compound is P-HPD (Fig. 1A, 4). Although all the assignments are consistent with the proposed identification, we could not detect the correlation between carbon-4 or carbon-2 and proton-3 (supplemental Fig. S1), preventing us from firmly linking both ends of this compound. Importantly, all the carbon resonances assigned to P-HPD are very similar to P-TPO differing only in carbon 4, which is a ketone group (with a carbon resonance at 208.4 ppm) in P-HPD and a gem-diol (carbon resonance at 98.6 ppm) in P-TPO. Thus, P-TPO is the hydrated form of P-HPD position C4.

Production of P-TPO, 3, from P-DPD, 2, can be easily explained by the interconversion of the ketone at position C3 and the hydroxide at C4 from P-DPD, 2, via the formation of 3,4-enediol intermediate to give the isomeric hydroxyketone, P-HPD, 4, and then hydration at position C4 (Fig. 1A). The proposed keto-enol tautomerization is common in well charac-
terized biochemical pathways, and such interconversion can occur spontaneously with any general α-hydroxyketone but is usually slow in the absence of a suitable catalyst (38). Furthermore, such ketone groups are often in equilibrium with their hydrated forms (39). This is also consistent with our NMR data and strengthens the conclusion that the minor product of the LsrG reaction is P-HPD, the non-hydrated form of P-TPO in Fig. 1A. In this particular case the hydrated molecule is the predominant form. Thus, based on the proposed reaction for formation of P-TPO (summarized in Fig. 1A), we conclude that LsrG catalyzes the isomerization of P-DPD.

P-TPO Is Highly Susceptible to Spontaneous Oxidation—When the LsrG/P-DPD reaction is performed in aerobic conditions and the absence of reducing agents, the products observed are P-TPO, P-HPD, P-TetraPO, phosphoglycolate, and other minor compounds (proton and phosphorous spectra, supplemental Fig. S3 and S4, respectively). The product P-TetraPO was assigned using a combination of proton, carbon, and phosphorous correlation NMR techniques as explained above (resonances are listed in Table 3). Phosphoglycolate had been previously assigned (18). The low concentrations of the other compounds precluded reliable identification by NMR.

Our results further showed that P-TetraPO and the other compounds mentioned above are produced from oxidation of P-TPO and P-HPD (supplemental Fig. S3, panel B). To determine whether this reaction is catalyzed by LsrG, we took advantage of the fact that diethylpyrocarbonate completely inhibits LsrG activity (data not shown). We measured the conversion of P-TPO and P-HPD into P-TetraPO in the presence of diethylpyrocarbonate or in its absence after bubbling oxygen through these samples. The decrease of P-TPO and P-HPD and the increase of P-TetraPO was very similar in both cases (supplemental Fig. S5), and thus we concluded that production of P-TetraPO does not require LsrG activity. It is also important to note that phosphoglycolate production was increased in the presence of oxygen; this could be due to the breakdown of P-TetraPO or the breakdown of P-TPO and P-HPD, but we have not pursued this issue. Additionally, as reported before (18) phosphoglycolate is also formed from P-DPD in the absence of LsrG, presumably as a result spontaneous degradation of P-DPD. We never observed the formation of P-TPO or P-HPD in the absence of LsrG. These data corroborate the conclusion that LsrG catalyzes the isomerization of P-DPD to P-TPO, which is in equilibrium with its non-hydrated form,
P-HPD, and that in the presence of oxygen these compounds are oxidized non-enzymatically to produce P-TetraPO, phosphoglycolate, and other minor compounds.

**Crystal Structure of LsrG and Implications for Catalysis** — To investigate the catalytic mechanism of LsrG, we determined the crystal structure of *E. coli* LsrG to 1.8 Å resolution. The crystal structure revealed that LsrG is a dimeric barrel (Fig. 4A). After building the protein structure, it became apparent that a large mass of non-protein electron density was present in a groove between the α-helices and β-strands of each subunit. We were unable to interpret this density either manually or using the ligand identification module in Phenix (26). Nonetheless, the location of this non-protein electron density suggests a possible location for the LsrG active site (Fig. 4B). Examination of this putative active site reveals several potential catalytic residues with side chains adjacent to the non-protein density, including Asn-25, Glu-54, and His-65 (Fig. 4C). His-70 might also be significant, acting to position His-65 properly.

A DALI search of the Protein Data Bank identified several close structural homologues including YgiN (1.7 Å r.m.s.d., 16% ID), SnoaB (1.8 Å r.m.s.d., 18% ID), ActVA-Orf6 (2.3 Å r.m.s.d., 17% ID), and IsdG (2.5 Å r.m.s.d., 16% ID) as well as a variety of hypothetical or uncharacterized proteins (40). All of these homologues have monooxygenase activity, but they act on a variety of substrates and in a range of pathways from antibiotic synthesis to heme degradation (for review, see Ref. 20). Interestingly, when LsrG is structurally aligned with these proteins, the observed non-protein density closely aligns with the active sites of each, further supporting the assignment of this region as the active site in LsrG.

Examination of the active sites shows very little conservation of catalytic residues between LsrG and its structural homologues. Most notably, a tryptophan residue (Trp-66 in ActVA-Orf6) that is conserved in the other homologues is replaced by a tyrosine (Tyr-56) in LsrG. In the monooxygenases this conserved tryptophan has been implicated in substrate specificity (ActVA-Orf6 (21), YgiN (41), SnoaB (42)) and may play a role in catalysis; the fact it is not conserved in LsrG suggests a significantly different substrate or, potentially, a different functionality for this enzyme. Other active site residues from these proteins are not generally conserved in LsrG, although there is modest conservation with YgiN, the closest homologue (41). Of the residues that Adams and Jia (41) suggest could play a catalytic role in YgiN, only His-75 is structurally conserved (His-65 in LsrG). His-65 is one of the residues we identified as potentially catalytic in LsrG (above); one of the others, Glu-54, is also conserved in YgiN (Glu-64).

**Sequence Comparison with LsrG from Other Organisms** — In earlier work (12) we used bioinformatics analysis to identify bacterial species that have a functional *lsr* operon. Using the sequences for LsrG from these species, we created a multiple sequence alignment via ClustalW2 (29). Only nine residues are completely conserved across all species (in addition to the start Met), and of these, only two residues (Gly-34 and Asp-39) are not positioned near the putative active site. These residues are likely conserved for structural reasons; Asp-39, in particular, is positioned near the dimer interface and forms hydrogen bonds with Arg-37 in the other chain.

Of the other seven residues, two (Glu-54 and His-65) are directly adjacent to the putative ligand density as shown in Fig.
4C and mentioned above. The position of these residues coupled with the fact that they are conserved across species suggests that they are likely to play catalytic roles. The remaining residues are in or around the active site and may play roles in determining the geometry of the active site. His-70 is positioned to form a hydrogen bond with Glu-32, which in turn forms a hydrogen bond with His-65. Val-75 is directly adjacent to the putative ligand density but, given its hydrophobic nature, is unlikely to be directly catalytic. Phe-18 may be involved in a stacking interaction with two other Phe residues, one of which (Phe-22) is directly adjacent to the active site. Asn-35 is adjacent to Tyr-56, a residue notable for occupying the position of the tryptophan residue widely conserved in the monoxygenases. Thus, the complete conservation of putative catalytic residues and the positioning of the other conserved residues support our identification of the LsrG active site.

**Site-directed Mutagenesis of the LsrG and Impact on Enzyme Activity and Activation of lsr Transcription**—Based on the location of the putative active site in the crystal structure and the analysis of sequence alignments (above), we identified four residues (Glu-54, His-65, Asn-25, and His-70) most likely to play a major role in catalysis. To test these four residues for a catalytic role, we generated individual alanine substitution mutants and tested these mutants (E54A, H65A, N25A, and H70A) for enzymatic activity in vitro using purified protein and in vivo via LsrG expressed from multicopy plasmids in an lsrG deletion mutant. As shown in Table 4, mutants E54A, H65A, and H70A had no or very low activity in vitro, and the N25A mutant had a significant reduced activity (about 3-fold less activity than WT). For the in vivo assays, we used a strain with a lsr-lacZ transcriptional reporter fusion because transcription of the lsr operon is activated by P-DPD (14, 17). The WT strain has lower lsr-lacZ activity than the lsrG deletion mutant (two first bars, Fig. 5) presumably because the absence of LsrG results in P-DPD accumulation. Accordingly, complementing the mutant with an isopropyl 1-thio-β-D-galactopyranoside-inducible multicopy plasmid expressing the lsrG-WT gene caused a reduction in lsr-lacZ activity (third bar, Fig. 5). We compared the ability of the plasmids expressing LsrG-WT to induce lsr transcription with that of plasmids expressing the single point mutants. In agreement with the in vitro results, ectopic expression from the plasmids containing E54A, H65A, or H70A lsrG did not complement the effect of the lsrG deletion on lsr transcription. However, complementation with plasmid carrying the N25A mutant was as effective as with plasmid containing LsrG-WT (Fig. 5). This correlates with in vitro result where N25A shows detectable activity. Although the activity in vitro is still lower than in WT, this difference is not observed in our in vivo system. These data show that the N25A mutant is slightly catalytically impaired, suggesting that Asn-25 plays a role in catalysis but that the catalytic defect can be overcome by expression from a multi-copy isopropyl 1-thio-β-D-galactopyranoside-inducible plasmid with the resulting higher concentration of protein in the cell.

**Impact of LsrG on Intracellular AI-2 P Accumulation**—Previous work showed that the lsr operon is mainly induced by P-DPD, and thus the fact that expression of the lsr operon in an lsrG mutant is higher than in the WT leads to the suggestion that the mutant accumulates more P-DPD than the WT (17). If this is the case, overexpression of plasmid-borne LsrG in this mutant would prevent this accumulation. To test this hypothesis, we measured the accumulation of P-DPD in cell extracts of the following E. coli strains: WT, the lsrG deletion mutant, and the mutant complemented with pLsrG-WT. The levels of free P-DPD were determined by proton NMR of extracts from cell suspensions actively metabolizing DPD. P-DPD was easily detected in the lsrG deletion mutant (P-DPD = 0.16 mM) but not in the other extracts (Fig. 6). Apparently, the rate of turnover of P-DPD in the strain overexpressing pLsrG-WT and even in the WT strain is sufficiently high that the P-DPD level was below the detection limits (i.e. lower than 0.01 mM). As a negative control we assayed cell extracts from the lsrG mutant when no DPD was added to the cell suspension, and as expected, no resonance with the P-DPD chemical shift was detected (Fig. 6, lower trace). A mutant in the kinase that phosphorylates DPD (lrSK) was used as an additional negative control, and again, no P-DPD was detected (Fig. 6). This direct measurement of P-DPD levels supports the earlier prediction that the increase in lsr transcription observed in lsrG mutants is a result of P-DPD accumulation.

**TABLE 4**

Specific enzyme activities of LsrG-WT and single amino acid mutants: E54A, H65A, N25A, and H70A

<table>
<thead>
<tr>
<th>Protein</th>
<th>Units/mg of protein&lt;sup&gt;a&lt;/sup&gt;</th>
</tr>
</thead>
<tbody>
<tr>
<td>LsrG-WT</td>
<td>61.0 ± 2.2</td>
</tr>
<tr>
<td>LsrG-E54A</td>
<td>&lt; 0.1</td>
</tr>
<tr>
<td>LsrG-H65A</td>
<td>&lt; 0.1</td>
</tr>
<tr>
<td>LsrG-N25A</td>
<td>20.4 ± 0.8</td>
</tr>
<tr>
<td>LsrG-H70A</td>
<td>0.8 ± 0.1</td>
</tr>
</tbody>
</table>

<sup>a</sup> One unit of enzyme activity is defined as 1 μmol of P-DPD consumed/min at 30°C.

![FIGURE 5. The effect of lsrG and its mutants on the lsr expression in E. coli. β-Galactosidase activity of the lsr-lacZ transcription reporter fusion was measured in WT and ΔlsrG strains expressing pProExH7b (empty vector). Activity was also measured in ΔlsrG strains carrying the pProExH7b with wild type lsrG (lsrG-WT) or the lsrG single amino acid alanine substitutions lsrG mutants E54A, H65A, N25A, and H70A. The error bars indicate S.D.](http://www.jbc.org/)

**Note:** All procedures were performed as previously described (14, 17). A list of the primers used is provided in the Supporting Information.
Isomerization of P-DPD by LsrG

**Figure 6. P-DPD accumulation in E. coli cell suspensions metabolizing DPD.** The methyl region of the $^1$H NMR spectra of cell extracts from different E. coli strains is shown. Each trace shows the spectrum of the extract obtained from strains: JC53 (WT + pProExHTb), JC47 (ΔlsrG + pProExHTb), JC43 (ΔlsrG + pIsr-G-WT), and JC62 (ΔlsrK + pProExHTb) from top to bottom. Extracts were obtained from cell suspensions incubated at 30 °C for 15 min with 2 mM DPD (except for the lower trace (strain JC47) to which no DPD was added). Cell extracts were performed as described under "Experimental Procedures." The relevant genotype of the strains used in each case is indicated next to the trace. The resonances of the methyl groups from P-DPD and the three forms of DPD are highlighted.

**DISCUSSION**

P-DPD is the intracellular inducer of the lsr operon, the gene products of which form a system capable of terminating AI-2-mediated bacterial behaviors (11, 43). Using NMR, x-ray crystallography, and a combination of *in vitro* and *in vivo* assays, we showed that LsrG catalyzes the isomerization of P-DPD to P-TPO (and its non-hydrated form, P-HPD). Our finding was surprising given our previous report (18), but it is not inconsistent. We had suggested that LsrG catalyzes the cleavage of P-DPD to phosphoglycolate and other unknown products. The experiments reported here show that the increased formation of phosphoglycolate observed in the presence of LsrG occurs only when the products of P-DPD isomerization are exposed to oxidative conditions. In the presence of oxygen, P-TPO and P-HPD are oxidized to P-TetraPO, phosphoglycolate, and other minor compounds, but this oxidation does not require active LsrG. Under anaerobic conditions we never observed formation of P-TetraPO or phosphoglycolate. Additionally, we also never observe these two compounds when the reaction is performed in the presence of the reducing agents DTT or glutathione (supplemental Figs. S6 and S7). Overall, our results indicate that the physiological function of LsrG is to catalyze P-DPD isomerization. Whether or not the products of this reaction are oxidized to phosphoglycolate *in vivo* will depend on the local oxidizing/reducing conditions and, potentially, the efficiency of other enzymes that act on the P-DPD isomers, and this is a question we are pursuing through ongoing experiments. To completely reveal the metabolic fate of DPD, it is necessary to identify and characterize the next steps of this pathway. The answer will provide clues to other benefits the bacteria may derive from the processing of DPD, metabolic or otherwise.

Using NMR to measure the accumulation of P-DPD in extracts of cells actively metabolizing DPD, we showed that the lsrG deletion mutant accumulates at least 10 times more P-DPD than WT. This assay provides a novel and direct way to measure intracellular P-DPD and can be applied to the study of other enzymes involved in P-DPD processing or regulators of this pathway. In this particular case, this observation provided direct evidence on the impact of *lsrG* in *vivo* and explains the high levels of Lsr expression observed in this mutant. Our results also showed that the lsrG deletion mutant has an aberrant profile of AI-2 internalization. Thus, LsrG is essential for the proper functioning of the Lsr system in interfering with AI-2-mediated signaling.

The crystal structure of P-DPD showed that LsrG has the same fold as enzymes belonging to the cofactor-independent monooxygenase family such as ActVA-Orf6, YgiN, and SnoaB (21, 41, 42). However, our data indicate that, unlike the other enzymes from this family, LsrG does not catalyze the incorporation of oxygen into the substrate; thus, LsrG plays a different catalytic role than other members of the family characterized so far. It is notable, however, that the keto-enol tautomerization step we propose for the LsrG reaction is also a step in the pathway of ActVA-Orf. Thus, the isomerization catalyzed by LsrG could be setting the stage for the oxidation reaction that leads to the observed production of P-TetraPO and phosphoglycolate.
Whether this oxidation occurs in vivo remains an open question.

Identification of the crystal structure led us to identify potential catalytic residues, and we assayed mutants of these sites. Our results showed that the residues Glu-54, His-65, and His-70 are essential for catalysis in vitro and that alanine substitutions of these residues are impaired in complementing the \( \text{lsrG} \) deletion mutant in vivo. These catalytic residues from LsrG are not structurally conserved in other proteins from the monooxygenase family studied so far. The same is largely true between the catalytic residues of other members of this family and can be explained by divergent adaptation to a variety of substrates (20, 42). One exception is the residue His-65, which is structurally conserved in YgiN (His-75), the closest structural homologue to LsrG, and demonstrated to be important for catalysis in both LsrG and YgiN (41).

Although the conservation of catalytic residues among the proteins from this family is low, this is not the case among the previously identified LsrG orthologs (12). In this group there was 100% conservation of 7 residues in and around the LsrG deletion mutant in vivo. These catalytic residues from LsrG are not structurally conserved in other proteins from the monooxygenase family studied so far. The same is largely true between the catalytic residues of other members of this family and can be explained by divergent adaptation to a variety of substrates (20, 42). One exception is the residue His-65, which is structurally conserved in YgiN (His-75), the closest structural homologue to LsrG, and demonstrated to be important for catalysis in both LsrG and YgiN (41).

Although the conservation of catalytic residues among the proteins from this family is low, this is not the case among the previously identified LsrG orthologs (12). In this group there was 100% conservation of 7 residues in and around the active site (which include the residues Glu-54 and His-65 and His-70) out of a total of only 9 completely conserved residues (in addition to the start Met). Asn-25, which is close to the potential active site, is only conserved in 17% of these proteins, suggesting that it plays a non-essential role in support of LsrG catalysis. This is in line with experimental results showing...

FIGURE 7. Profile of extracellular AI-2 accumulation and expression of \( \text{lsr} \) during growth. Extracellular AI-2 (A) and \( \beta \)-galactosidase activity of the \( \text{lsr-lacZ} \) transcription reporter fusion (B) were measured in WT (MG1655, circles), \( \Delta \text{lsrK} \) (KX1186, squares), and \( \Delta \text{lsrG} \) (JCM23, triangles) *E. coli* strains. Measurements were performed on aliquots collected at indicated times. The error bars indicate S.D.
Isomerization of P-DPD by LsrG

reduced activity for the N25A mutant in vitro (though not in vivo, see above). The conservation of other active site residues, including those shown here to be essential for function, supports the prediction that the function of LsrG in the regulation of the Lsr system is conserved in the range of species with LsrG homologues. The lsr operon and lsrG homologues are present in bacteria including important human pathogens such as S. typhimurium, B. anthracis, and Y. pestis (12) and thus can impact communications in multispecies bacterial communities that include pathogens such as these.

The belief in the field is that quorum sensing can be exploited to manipulate bacterial behavior and thus lead to an alternative to traditional antibiotics. This is mainly because therapies that interfere with quorum sensing are predicted to create lower selective pressure toward resistance than traditional antibiotic treatments (44, 45). A variety of approaches are being developed that interfere at different steps on the quorum-sensing pathways. One of these strategies, referred to as quorum quenching, involves altering or destroying the autoinducer signal (46). Most reports on quorum-quenching approaches have employed naturally occurring enzymes that degrade species-specific signals. Enzymes that degrade acyl homoserine lactones (AHLs) include AHL lactonases (47), AHL acylases (48), and paraoxonases (49). Fewer studies have investigated interference with autoinducer peptides, but examples of quorum-sensing inhibition in Staphylococcus aureus are reviewed in Lyon and Novick (50). In contrast, the Lsr system and LsrG in particular targets the interspecies signal AI-2. This is the only natural system known so far to degrade AI-2 signal, and there are multiple studies showing that the Lsr system can function as an AI-2 quenching system (11, 43). Our work shows that interference with LsrG alters the function of Lsr, suggesting that it could be a useful target for AI-2-based quorum-quenching therapies. Interference with interspecies cell-cell signaling can be particularly important when treating infections caused by multispecies biofilms that might rely on interspecies signals for persistence. For these cases, we predict that targeting interspecies signaling is likely to be more effective than targeting species-specific signals.

Acknowledgments—We thank Frederick M. Hughson (Princeton University) and Helena Santos (Instituto Tecnologia Quimica e Biologica) for critically reading the manuscript. We thank Ana R. Neves and Teresa Catarino for precious help in optimizing the NMR conditions and preparing the anaerobic samples, respectively. The NMR spectrometers are part of The National NMR Network (REDE/1517/RMN/2005), supported by “Programa Operacional Ciência e Inovação (POCTI) 2010” and Fundação para a Ciência e Tecnologia. We acknowledge National BioResource Project (Japan):E. coli for providing strain JC62.

REFERENCES

18342 JOURNAL OF BIOLOGICAL CHEMISTRY VOLUME 286 • NUMBER 20 • MAY 20, 2011

**Isomerization of P-DPD by LsrG**
Processing the Interspecies Quorum-sensing Signal Autoinducer-2 (AI-2): CHARACTERIZATION OF PHOSPHO-(S)-4,5-DIHYDROXY-2,3-PENTANEDIONE ISOMERIZATION BY LsrG PROTEIN

João C. Marques, Pedro Lamosa, Caitlin Russell, Rita Ventura, Christopher Maycock, Martin F. Semmelhack, Stephen T. Miller and Karina B. Xavier

doi: 10.1074/jbc.M111.230227 originally published online March 30, 2011

Access the most updated version of this article at doi: 10.1074/jbc.M111.230227

Alerts:
- When this article is cited
- When a correction for this article is posted

Click here to choose from all of JBC's e-mail alerts

Supplemental material:
http://www.jbc.org/content/suppl/2011/03/30/M111.230227.DC1

This article cites 45 references, 10 of which can be accessed free at
http://www.jbc.org/content/286/20/18331.full.html#ref-list-1